## B.Sc. II Semester Degree Examination, June - 2018 CHEMISTRY

Paper No : II

(New)

Time: 3 Hours

Maximum Marks: 80

## Instructions to Candidates:

The question paper contains 4 sections and answer all the 4 sections A, B, C and D as per instructions.

## Section - A

Answer any **TEN** of the following:

 $(10 \times 2 = 20)$ 

- 1. Give any two uses of S-block elements.
- What are inter halogen compounds? Give a preparation of TF<sub>5</sub>.
- Define bond order.

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- 4. Write the structure of diborane.
- 5. Calculate the angle strain in cyclopentane.
- **6.** Why alkynes are acidic in nature?
- 7. What are cumulated dienes? Give one example.
- 8. State Huckel's rule.
- 9. Define order & molecularity of a reaction.
- 10. What are parallel reaction give an example?
- 11. What are liquid crystals? Give an example.
- 12. Give any two applications of colloids.



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		Section - B	
	Ans	swer any two of the following:	(2×10=20)
13.	a)	Discuss the corparative study of alkali and alkaline earth metals with resmelting points and boiling points and flame colouration.	pect to density (6)
	b)	Write the preparation and structure of IF <sub>3</sub> and IF <sub>7</sub> .	(4)
14.	a)	According to USEPR theory explain the structures of BF <sub>3</sub> and PCl <sub>5</sub>	(6)
	b)	Write a note on Vander Waal's interactrons.	. (4)
15.	a)	Discuss the structure and bonding in XeF6 and Xeo3	(6)
	b)	Write a note on characteristics properties of oxides and hydroxiels of al	kalimetals.(4)
		Section - C	
	Ans	swer any TWO of the following:	(2×10=20)
16.	a)	How are alkanes synthesized by	
		i) Kolbe's reaction	
		ii) Corey - House reaction.	(6)
	b)	Explain the orientation effect of nitrobenzene.	(4)
17.	a)	Give any two methods of synthesis of alkynes.	(6)
	b)	Write a note on Sache - Mohr theory of strain less rings.	(4)
18.	a)	Give an account of Buyer's strain theory.	(6)
	b)	Explain conjugated and cumulative diches with examples give a method of 1,3 - butadiene.	ls of synthesis (4)
		Section - D	
	Ans	wer any <b>two</b> of the following:	(2×10=20)
19.	a)	Derive an expression for the velocity constant of a second order reaction	n when <i>a ≠ b</i> . <b>(6)</b>
	b)	Write a note on parallel and consecutive reaction.	(4)

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20.	a)	Discuss briefly intermolecular forces in liquids.	(6)
	b)	Write a note on electro osmosis.	(4)
21.	a)	Give the determination of order of a reaction by	(6)
		i) Differential rate equation method	
	Ť	ii) Graphical method.	
	b)	Give the structural differences between solids, liquids and gases.	(4)

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